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# **Research Article**

# Stability Indicating HPLC Method for the Simultaneous Estimation of Triamcinolone Acetonide and Benzyl Alcohol in Pure Form and Epirelefan<sup>®</sup> Vial

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# ABSTRACT

A rapid, sensitive, and accurate stability indicating HPLC method was developed for the simultaneous determination of triamcinolone acetonide and benzyl alcohol in pure form, degradation products and pharmaceutical preparation. The separation was carried out on RP BDS Hypersil<sup>®</sup> C<sub>18</sub> column (150 x 4.60 mm, 5µm) using an isocratic mobile phase composed of acetonitrile: 0.05 M phosphate buffer PH 3.50 (55 : 45). Both benzyl alcohol and triamcinolone acetonide quickly eluted at 1.67 min and 3.42 min, respectively, with a flow rate of 1.50 mL /min and UV detection at 254 nm. The linearity was in the range of 1 – 50 µg/mL for triamcinolone acetonide and 2 - 10 µg/mL for benzyl alcohol. The method has been validated for linearity, accuracy, precision, specificity, limit of detection, limit of quantification, robustness, and ruggedness as per the ICH guidelines. Finally, the method was compared statistically with reference methods indicating that there is no significant difference between them in respect of precision and accuracy.

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# Introduction

Triamcinolone is chemically (1S,2S,4R,8S,9S,11S,12R,13S)-12-fluoro-11-hydroxy-8-(2 hydroxyacetyl)-6,6,9,13-tetramethyl-5,7dioxapentacyclo[10.8.0.02,9.04,8.013,18]icosa-14,17-dien-16-one (Figure 1). Triamcinolone acetonide (TAM) is the acetonide salt form of triamcinolone, a synthetic glucocorticoid with an immunosuppressive and anti-inflammatory activity [1] (Link1). TAM binds to specific cytosolic glucocorticoid receptors and subsequently interacts with glucocorticoid receptor response element on DNA and alters gene expression [2]. This results in an induction of the synthesis of certain anti-inflammatory proteins while inhibiting the synthesis of certain inflammatory mediators [3]. Consequently, an overall reduction in chronic inflammation and autoimmune reactions are accomplished [4]. Few chromatographic methods have been reported also for determination of TAM either alone or in combination with other drugs in different matrices [5-11].

On the other hand, benzyl alcohol (BZA) is chemically named as phenyl methanol (Figure 1) and it is an aqueous soluble preservative widely used in injectable pharmaceutical preparations as well as in cosmetic products [12]. Although toxic in neonates and infants, it is generally recognized as safe by the FDA at concentrations up to 5% in adults [13]. The effect of BZA present as preservative in bacteriostatic normal saline has recently come under investigation when administered via intraarticular injection. Patients who undergo MRI-based arthrography for standard clinical indications are commonly administered a contrast agent like gadolinium diluted in bacteriostatic saline (Link2) [14]. Few chromatographic methods have been reported also for determination of BZA in combination with other drugs [15-18].

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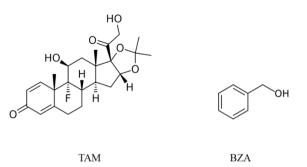


Figure 1: Chemical structures of triamcinolone acetonide (TAM) and benzyl alcohol (BZA).

To the best of our knowledge, this is the first paper dealing with stability indicating chromatographic separation of TAM and BZA despite they are formulated in a vial form. As such, the present work introduces a simple, rapid, reproducible, and sensitive HPLC method that has been established and validated for the determination of TAM & BZA mixture in their pure forms, degradation products and in their tablet dosage form according to International Conference on Harmonization (ICH) guidelines [19].

# Experimental

#### I Materials and Reagents

TAM and BZA were kindly supplied by EIPICO Pharmaceuticals Company, EGYPT. Epirelefan<sup>®</sup> vial (EIPICO) was labeled to contain 40 mg/mL TAM suspension in BZA. The used chemicals in all experiments were of analytical HPLC grade, methanol and acetonitrile (Lab Scan, Gliwice, PL), potassium dihydrogen phosphate (Fluka, Loughborough, UK), and orthophosphoric acid 85% (Merck, USA) were used. Distilled water was obtained through Millipore filter 0.45 µm throughout the whole experiment. A combination of acetonitrile : 0.05 M potassium hydrogen phosphate buffer pH 3.50 (55 : 45) was mixed and degaussed in a fast clean ultra sonicator for 5 minutes. Finally, the mixture was filtered through 0.45µm membrane filter. 0.1 N HCL, 0.1 N NaOH, and 0.35% H<sub>2</sub>O<sub>2</sub> were used for stability studies.

#### **II Instrumentation and Chromatographic Conditions**

WATERS<sup>®</sup> 2695 SEPARATION MODULE EIPICO Code: 0302QCI0112, equipped with Quaternary pump, Auto sampler, Variable wavelength detector and Column thermostat are operated by waters Empower 2 software. The pH of the mobile phase was adjusted using a Metrohm 744 pH meter (Herisau, Switzerland). RP-HPLC separation of TAM and BZA was achieved using RP BDS hypersil<sup>®</sup> C<sub>18</sub> column (150 x 4.60 mm, 5µm) (phenomenex, USA). The elution was isocratic and the mobile phase composed of 55% acetonitrile and 45% phosphate buffer (0.05 M potassium dihydrogen phosphate, adjusted to pH 3.50 by using orthophosphoric acid). The injection volume was 20 µL, and the temperature of column was kept at 40°C. Flow rate was adjusted at 1.50 mL/min and detection was achieved at 254 nm (Table 1). Prior to use, the mobile phase was degassed and filtered by passing through a 0.2 µm pore size membrane filter (Millipore, Milford, MA, USA).

Table 1: Chromatographic conditions for th	e proposed method.

Parameters	Conditions
Column	RP BDS Hypersil® C18 (150 x 4.60 mm,
Column	5µm) (phenomenex, USA)
Mahila ahaas	Acetonitrile : 0.05 M phosphate buffer pH
Mobile phase	3.50 (55 : 45)
UV detection, nm	254
Flow rate, mL/min	1.50
Injected volume, µl	20
Temperature	40

# **III Preparation of Standard Solutions**

An accurately weighed 10 mg of TAM standard and 10 mg of BZA standard were transferred to a 100 mL volumetric flask, dissolved and completed to volume with methanol.

# IV Construction of Calibration Curves

Aliquots of standard solution of TAM and BZA were taken and adjusted with methanol to obtain their final concentration in the range of 1-50  $\mu$ g/mL for TAM and 2- 10  $\mu$ g/mL for BZA. Calibration curves were constructed by plotting area under curve against the concentration.

# **V** Forced Degradation Samples Preparation

## i Acid Degradation

5 mL of the stock standard solution was transferred into 100 mL volumetric flask, then 10 mL of 0.1 N HCL was added and placed in a boiling water bath for about 3 minutes. The solution was cooled for 5 minutes, neutralized by addition of 0.01 N NaOH to pH 7 and completed to 100 mL with methanol.

#### ii Alkaline Degradation

5 mL of the stock standard solution was transferred into 100 mL volumetric flask, then 10 mL of 0.1 N NaOH was added and placed in a boiling water bath for about 3 minutes. The solution was cooled for 5 minutes, neutralized by addition of 0.01 N HCL to pH 7 and completed to 100 mL with methanol.

#### iii Oxidative Degradation

5 mL of the stock standard solution was transferred into 100 mL volumetric flask, then 5 mL diluted  $H_2O_2$  (5mL of 35%  $H_2O_2$  in 100 mL  $H_2O$ ) was added and completed to 100 mL with methanol.

#### iv Heat Degradation

5 mL of the stock standard solution was transferred into 100 mL volumetric flask, then 10 mL of distilled water was added and placed in a boiling water bath for about 3 minutes. The solution was cooled for 5 minutes and completed to 100 mL with methanol.

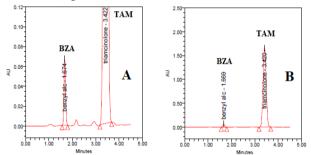
#### VI Preparation and Analysis of Pharmaceutical Preparations

2 ml of the well shaken Epirelefan<sup>®</sup> vial suspension (40 mg/mL TAM, 1 mg/mL BZA) was transferred into 100 mL volumetric flask, then the volume was completed to the mark with methanol. Working solutions were then prepared as in the above procedure and completed as previously mentioned under the general procedure.

# **Results and Discussion**

#### I Optimization of the Chromatographic Conditions

This method was optimized after studying many variables. Many columns were used to obtain the optimum separation and shortest run time. RP BDS hypersil<sup>®</sup> C<sub>18</sub> column (150 x 4.60 mm, 5 $\mu$ m) (phenomenex, USA) was used. In order to obtain sharp, separated peaks without tailing or distortion, further optimization was needed. As such, different wavelengths and percentages of mobile phase contents were used. Mobile phase composed of acetonitrile and 0.05 M phosphate buffer PH 3.50 (55 : 45) respectively gave the best results with wavelength at 254 nm. Under these conditions, BZA and TAM in pure form can be separated and eluted at 1.67 and 3.42 min, respectively, as illustrated in (Figure 2A).



**Figure 2:** HPLC chromatogram of mixture of triamcinolone acetonide (TAM) and benzyl alcohol (BZA) in **A**) pure form and **B**) their tablets using RP BDS Hypersil<sup>®</sup> C<sub>18</sub> (150 x 4.60 mm, 5 $\mu$ m) and a mobile phase of acetonitrile : 0.05 M phosphate buffer pH 3.50 (55 : 45, v/v). Other chromatographic conditions are stated in (Table 1).

# **II Method Validation**

Method Validation of the developed method was carried out according to ICH guidelines for its linearity, accuracy, precision, specificity, robustness, ruggedness, limit of detection and limit of quantification [19].

## i System Suitability Parameters

The column efficiency, resolution, and peak symmetry were calculated for the standard solutions (n=6). The values obtained demonstrated the suitability of the system for the analysis of this drug combination (Table 2). The optimum mobile phase showed symmetrical peaks (1.03 < T < 1.14), capacity factor (1 < k < 10), resolution > 2 and theoretical plates> 2000 as per required by the center for drug evaluation and research [20].

**Table 2:** System suitability parameters for triamcinolone acetonide (TAM) and benzyl alcohol (BZA) in their pure form.

Parameters	BZA	TAM Reference values [20]	
Retention time, tr	1.67	3.42	
Capacity factor, k'	1.09	3.27 Accepted k' value (1-1	10)
Peak asymmetry (Tailing	1.14	1.03 Accepted T value $\leq 2$	
factor, T)			
Therotical plates, N	2726	2869 Accepted N value > 20	)00
Resolution, Rs		7.61 Accepted value > 2	
Selectivity (Separation		3.00	
factor, α)			

# ii Linearity

Linearity of this method was studied by preparing five different concentrations of the standard solutions of TAM and BZA. Area under peak of these solutions were collected and the calibration curves were obtained by plotting absorbance vs. concentration of the drugs showing linearity in the concentration range of 1-50  $\mu$ g/mL for TAM and 2-10  $\mu$ g/mL for BZA (Table 3). Linear regression equations were found to be y = 8.545x + 0.3018, and y = 1615420x + 1269, while the regression coefficient values (r) were 1 and 0.9999 for TAM and BZA, respectively, indicating a high degree of linearity.

TAM			E	BZA	
Taken	Found	Recovery	Taken	Found	<b>Recovery %</b>
μg/mL	μg/mL	%	μg /mL	μg /mL	
1	0.99	99.09	2	1.99	99.73
5	4.98	99.53	4	4.03	100.63
10	10.05	100.53	6	6.01	100.08
20	19.98	99.91	8	7.94	99.19
50	49.84	99.67	10	10.04	100.40
Mean		99.75			100.01
±SD		0.53			0.02
±RSD		0.53			0.02
±SE		0.24			0.019
Variance		0.28			0.32
LOD (µg/mL)		0.19			0.15
LOQ (µg/mL)		0.62			0.46

#### iii Accuracy

Accuracy was determined by using concentrations in the calibrations range from the vial sample. These concentrations represent the lower part of calibration curve (5  $\mu$ g/mL of TAM and 4  $\mu$ g/mL of BZA), the

#### Table 4: Results of accuracy study (recovery).

middle part (10 & 20  $\mu$ g/mL of TAM and 6 & 8  $\mu$ g/mL of BZA), and the higher concentration part (50  $\mu$ g/mL of TAM and 10  $\mu$ g/mL of BZA). Mean recoveries were reported to be 99.83 and 100.85 while RSD were calculated to be 0.51 and 0.87 for TAM and BZA, respectively, indicating a high degree of the method accuracy (Table 4).

Sample No.		TAM			BZA	
	Nominal conc. (µg/mL)	Found conc. (µg/mL)	) Recovery %	Nominal conc. (µg/mL)	Found conc. (µg/mL)	<b>Recovery %</b>
1	5	4.96	99.30	4	4.04	101.13
2	10	10.05	100.49	6	6.03	100.57
3	20	19.98	99.88	8	8.15	101.89
4	50	49.83	99.6528	10	9.98	99.81
Mean			99.83			100.85
SD			0.50			0.88
RSD			0.51			0.87

#### iv Precision

Precision was calculated in terms of method precision and intermediate precision. Method precision (% repeatability) of the proposed HPLC method was determined by 6 replicate injections of different homogenous samples of TAM and BZA in the same day, while

intermediate precision (day to day precision) was determined by 3 replicate injections of TAM and BZA per day table. However, the % recovery was within the acceptable range of 98 to 102 % and RSD of these injections does not exceed 2% as reported in (Table 5), indicating that the developed method is highly precise and reproducible.

David	aana ug /mI	Intra-day		Inter-day	Inter-day	
Drug	conc. μg /mL	Mean ± SD	RSD	Mean ± SD	RSD	
ТАМ	10	$100.95 \pm 0.083$	0.0824	$101.02 \pm 0.078$	0.077	
BZA	4	$101.44\pm0.17$	0.17	$101.86\pm0.12$	0.12	

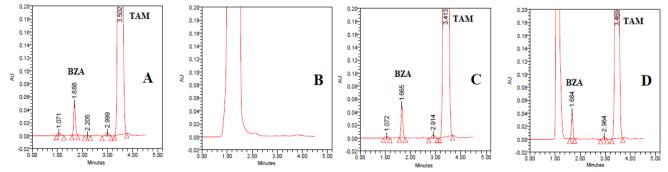
#### v Limit of Detection and Limit of Quantification

The limit of detection (LOD) was calculated from the linearity curve using the formula LOD =  $3.3 \times \text{SD/Slope}$  while the limit of Quantification (LOQ) was calculated from the linearity curve using the formula LOQ =  $10 \times \text{SD/Slope}$ . The LOD for TAM was confirmed to be 0.19 µg/mL and that for BZA was 0.15 µg/mL. On the other hand, The LOQ for TAM was calculated to be 0.62 µg/mL and 0.46 µg/mL for BZA (Table 3). These results indeed show that the proposed method is

highly sensitive and applicable for pharmaceutical studies even if detection of small concentrations in the nanogram range is required.

#### vi Specificity

The specificity of the analytical method was checked in whereas, the chromatograms of the vial or the degradation products under different drastic conditions illustrate that there is no interference of the peaks of excipients or the degradation products in the determination of TAM and BZA as seen in (Figures 2B & 3).



**Figure 3:** HPLC chromatogram of mixture of triamcinolone acetonide (TAM) and benzyl alcohol (BZA) in presence of **A**) acid degradation products, **B**) alkaline degradation products, **C**) heat degradation products, and **D**)  $H_2O_2$  degradation products using RP BDS Hypersil<sup>®</sup> C<sub>18</sub> (150 x 4.60 mm, 5µm) and a mobile phase of acetonitrile : 0.05 M phosphate buffer pH 3.50 (55 : 45, v/v). Other chromatographic conditions are stated in (Table 1).

# vii Ruggedness and Robustness

Ruggedness of the method was determined by carrying out the experiment on by different operators using the same chromatographic conditions. Robustness of the method was determined by subjecting the

#### Table 6: Results of robustness.

method to slight changes in the chromatographic conditions. No significant changes in the chromatograms were observed, where all SD values were less than 2 (Table 6), proving that the developed method is highly rugged and robust.

Drug	TAM	BZA
	Mean ± SD	Mean ± SD
Flow rate 1.50	$99.90\pm0.08$	$101.44 \pm 0.17$
Flow rate 1.55	$101.86 \pm 0.12$	$99.65 \pm 0.10$
Mobile phase 55:45	$99.90\pm0.08$	$101.44 \pm 0.17$
Mobile phase 54:46	$105.31 \pm 0.57$	$106.45 \pm 1.00$

# viii Analysis of Vial

EPIREFELAN<sup>®</sup> vial which contains 40 mg/mL TAM and 1 mg/mL BZA has been successfully analysed by the proposed RP-HPLC method. Excipients and impurities did not show interference indicating high specificity of the method. Results obtained were compared to those

obtained by applying reference methods where Student's t-test and Ftest were performed for comparison [5, 15]. Results shown in (Table 7) indicated that calculated t and F values were less than tabulated ones for TAM and BZA which in turn indicate that there is no significant difference between proposed method and reference ones relative to precision and accuracy.

Table 7: Statistical analysis of results obtained	by the proposed HPLC method	od applied on pharmaceutical formula	tion compared with reference methods.

	Recovery ± SD	Recovery ± SD					
	Proposed	Proposed			method	Student t- values	F- Values
	Method	Ν	Method	Ν	number	t- values	
TAM	$100.70\pm0.14$	4	$100.60\pm0.32$	4	[5]	<b>0.08</b> (1.94) <sup>a</sup>	<b>5.22</b> (9.28) <sup>b</sup>
BZA	$100.90\pm0.44$	4	$99.18\pm0.75$	4	[15]	<b>1.92</b> (1.94) <sup>a</sup>	<b>2.92</b> (9.28) <sup>b</sup>
a that		1 -					

 $^{\rm a}$  and  $^{\rm b}$  are the Theoretical Student t-values and F-ratios at p=0.05.

# Conclusion

In this study, an accurate, rapid, simple and sensitive RP-HPLC method with UV detection was developed for the assay of triamcinolone acetonide and benzyl alcohol in pure form, in presence of degradation products under different drastic conditions and in vial form. The complete separation of the analytes was achieved within short chromatographic run time (4 min) with no interfering peaks where there is no need for traditional HPLC methods with complex mobile phase mixtures, long chromatographic run times and more solvent consumed methods. The method was compared statistically with reference methods indicating that there is no significant difference between them in respect of precision and accuracy.

#### **Conflicts of Interest**

None.

## Ethical Approval

This manuscript does not include any studies on human or animals.

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